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(54) ORAL PREPARATIONS

COLGATE-PALMOLIVE COMPANY, a Corporation organised and existing under the Laws of the State of Delaware, United States of America, of 300 Park Avenue, New York, New York 10022, United States of America, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particu-10 larly described in and by the following statement:-

This invention relates to oral preparations, i.e. preparations suitable for application to the oral cavity, including, for example, denti-15 frices such as toothpastes or dental creams, tooth powders and liquid dentifrices, mouthwashes or rinses, dental chewing gums, dental floss, tablets, lozenges and troches.

According to the present invention an oral preparation comprises, as a non-toxic antibacterial agent, an amino-cyano polymeric diguanide or salt thereof as hereinafter set forth, and as a vehicle for the antibacterial agent, a mixture of water and alcohol or a solid water-insoluble dental polishing material.

The said amino-cyano polymeric diguanide is a polymer or a mixture of polymers having the formula:-

30 wherein Y represents

or

$$N \equiv C - NH - C - NH - (CH_2)_5 - MH$$

[Price 25p]

---NH₂

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and n is an integer from 2 to 6. The average value of n in the mixture of polymers may be about 4. A salt or mixture of salts of such polymer or polymers may also be employed.

The said amino-cyano polymeric diguanides are cationic but compatible with anionic substances and other materials such as are commonly present in oral preparations, for example, surface active agents, fluorine-providing compounds and, when the oral preparations such as a dentrifice contain a sparingly soluble polishing material, that portion of the polishing material which is water-soluble or saliva-soluble.

In one form of the invention the aminocyano polymeric diguanide employed is prepared in accordance with a procedure substantially as set forth in British Patent Specification No. 1,152,243. Such a procedure comprises reacting hexamethylene diamine,

H₂N(CH₂)₀NH₂,

or an inorganic salt thereof with the hexamethylene diamine salt of a dicyanimide having the formula

$[H_3N-(CH_2)_6-NH_3][N(CN)_2].$

At least the initial part of the reaction typically takes place in the presence of a hydroxylic solvent such as an aliphatic alcohol containing 1 to 4 carbon atoms, or more preferably water. The hydroxylic solvent may be mixed with a minor amount of a non-hydroxylic solvent, such as toluene. Preferably, the pH of the reaction mixture is adjusted to 6 to 8

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with a mineral acid, such as hydrochloric acid, which converts the free diamine to the acid salt, and the solution is initially heated to about 100°C at atmospheric pressure to distil off at least part of the hydroxylic solvent. The reaction mixture is then heated, e.g. to a temperature in the range from 120° to 170°C, for 1 to 8 hours. Then, while cooling, water may be added if desired. The thusobtained amino-cyano polydiguanide can then be separated from any water-immiscible solvent present and, if desired, also isolated from the aqueous solution.

The average molecular weight of the poly-15 mers produced by the above procedure is

generally about 1100.

As already indicated, non-toxic addition salts of such polymers may be employed. Such salts may be inorganic, e.g. fluoride, chloride, nitrate or sulphate, or organic such as succinate, tartrate, citrate, maleate and methosulphate.

Certain of the addition salts are insoluble, such as the water-insoluble sulphate addition salt. These salts may be incorporated into dentifrice compositions as discrete particles which may be visible on the surface of opaque dental creams or within and on the surface of visually clear, i.e. transparent or translucent, gels.

The preferred polymeric product is a mixture of various combinations of the X and Y members and various amounts of the recur-

ring unit

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Typically the polymeric product is prepared as the chloride salt and appears to comprise mainly the linear amino-cyano polyhexamethylene diguanide:

This product is the polymeric diguanide used in the Examples.

The polymer is very soluble in water and sailva and is characterized by a high degree of antibacterial activity comparable to that of 1,6 - di - (p - chlorophenylbiguanido - hexane).

The polymer may be employed generally in amounts in the range from 0.01% to 5%, preferably 0.05% to 1.5%, by weight of the oral preparation.

In certain forms of the invention the oral preparation may be substantially liquid in character, such as a mouth wash or rinse. In

such a preparation the vehicle is a water-alcohol mixture. Generally, the ratio of water to alcohol is in the range of 1:1 to 20:1, preferably 3:1 to 20:1 and most preferably about 17:3, by weight. The total amount of water-alcohol in this type of preparation is typically in the range from 70% to 99.95% by weight of the preparation. The pH of such liquid preparations is generally in the range from 4.5 to 9, typically from 5.5 to 8.

Such liquid oral preparations may also contain a surface active agent and/or a fluorine-

providing compound.

In certain other forms of this invention, the oral preparation may be substantially solid or pasty in character, such as a tooth powder, or a toothpaste or dental cream. The dental vehicle of such solid or pasty oral preparations may comprise a solid water-insoluble polishing material. Examples of polishing materials are water-insoluble sodium metaphosphate, potassium metaphosphate, tricalcium phosphosphate, phate, dihydrated dicalcium anhydrous dicalcium phosphate, calcium pyrophosphate, magnesium orthophosphate, trimagnesium phosphate, calcium carbonate, alumina, aluminium silicate, zirconium silicates, silica, plastics such as polymethyl methacrylate, bentonite and mixtures thereof. Preferred polishing material include crystalline silica having particles of sizes of up to 5 microns, a mean particle size of up to 1.1 microns and a surface area of up to 50,000 cm2/gm, insoluble sodium metaphosphate, anhydrous dicalcium phosphate and calcium carbonate which typically are finely divided to have an average particle size below 10 microns.

Alumina, particularly the hydrated alumina sold by Alcoa (United States of America) as "C333" which has an alumina content of 64.9% by weight, a silica content of 0.008% by weight, a ferric oxide content of 0.003% and a moisture content of 0.37% at 110°C and which has a specific gravity of 2.42 and a particle size such that 100% of the particles are less than 20 microns, is particularly desir-

able.

When visually clear gcls are employed, a polishing agent of colloidal silica, such as those sold under the Trade Mark SYLOID as "Syloid 72" and "Syloid 74" or under the Trade Mark SANTOCEL as "Santocel 100" and alkali metal aluminosilicate complexes are particularly useful, since they have refractive indices close to the refractive indices of gelling agent-liquid (including water and/or humectant) systems commonly used in dentifrices.

Many of the so-called "water-insoluble" polishing materials are anionic in character and also include small amounts of soluble material. Thus insoluble sodium metaphosphate may be formed in any suitable manner as illustrated by Thorpe's Dictionary of Applied Chemistry, Vol. 9 (4th ed.), pp.

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510—511. The forms of insoluble sodium metaphosphate known as Mandrell's salt and Kurrol's salt are further examples of suitable materials. These metaphosphate salts exhibit a minute solubility in water, and therefore are commonly referred to as insoluble metaphosphates. There is present therein a minor amount of soluble phosphate material as impurities, usually a few percent such as up to 4% by weight. The amount of soluble phosphate material, which is believed to be a soluble sodium trimetaphosphate in the case of insoluble sodium metaphosphate in the case of insoluble alkali metal metaphosphate is typically employed in powder form of a particle size such that no more than 1% of the material is larger than 37 microns.

The polishing material is generally present in amount in the range from 20% to 99% by weight of the oral preparation. Preferably, it is present in amount in the range from 20% to 75% in toothpaste and in the range from 70% to 99% in tooth powder.

In the preparation of tooth powders, it is usually sufficient to admix mechanically, e.g., by milling, the various solid ingredients in appropriate quantities and particle sizes.

In pasty oral preparations the antibacterial diguanide polymer is compatible with the polishing material and other components of the preparation. Thus, in a toothpaste, the liquid vehicle may comprise water, typically in amount in the range from 10% to 90% by weight of the preparation. Glycerine, sorbitol or propylene glycol may also be present as humectants or binders. Particularly advantageous liquid ingredients comprise mixtures of water, glycerine and sorbitol. In clear gels where the refractive index is an important consideration, 10-30% by weight of water, 0 to 80% by weight of glycerine and 20-80% by weight of sorbitol is preferably employed. A gelling agent, such as natural or synthetic gums or gum-like materials, typically Irish moss, sodium carboxymethylcellulose, methyl cellulose, or hydroxyethyl cellulose, may be employed. Other gelling agents which may be employed include synthetic inorganic silicated clay sold under the Trade Mark LAPONITE as "Laponite CP" and "Laponite SP", having the formula

[Si₈Mg_{5,1}Li_{9,6}H_{7,6}O₂₄]^{0,6}-Na+_{0,63}

gum tragacanth, polyvinylpyrrolidone and starch. They are usually present in toothpaste in an amount up to 10% by weight, preferably in the range from 0.5% to 5%. The preferred gelling agents are methylcellulose, hydroxyethyl cellulose, "Laponite CP" and most preferably "Laponite SP". In a toothpaste or gel, the liquids and solids are proportioned to form a creamy or gelled mass which is extrudable from a pressurized con-

tainer or from a collapsible, e.g. aluminium or lead, tube.

The solid or pasty oral preparation may also contain a surface active agent and/or a fluorine-providing compound.

Surface active agents which may be employed are organic materials and afford increased prophylacetic action, and assist in achieving thorough and complete dispersion of the preparation throughout the oral cavity. The organic surface active material may be anionic, non-ionic, ampholytic or cationic in nature, and it is preferred to employ as the surface active agent a detersive material which imparts to the preparation detersive and foaming properties. Suitable such detergents are water-soluble salts of higher (C₁₂—C₁₈) fatty acid monoglyceride monosulphates, such as the sodium salt of the monosulphated monoglyceride of hydrogenated coconut oil fatty acids; higher alkyl sulphates, such as sodium lauryl sulphate; alkyl aryl sulphonates, such as sodium dodecyl benzene sulphonate; higher alkyl sulphoacetates; higher fatty acid esters of 1,2-dihydroxy propane sulphonates; and substantially saturated higher aliphatic acyl amides of lower (up to C₄) aliphatic amino carboxylic acids, such a those having 12 to 16 carbons in the fatty acid, alkyl or acyl radicals. Examples of the last mentioned amides are N-lauroyl sarcosine, and the sodium, potassium, and ethanolamide salts of N-lauroyl, N-myristyl or N-palmitoyl sarcosine. These should be substantially free from soap or similar higher fatty acid material which tends to reduce substantially the effect of these compounds. The use of these sarcosinate compounds in dentifrice preparations of the invention is particularly advantageous since these materials exhibit a prolonged and marked effect in the inhibition of acid formation in the oral cavity due to carbohydrate breakdown, in addition to exerting some reduction in the solubility of tooth enamel in acid solu-

Nonionic surface active agents include condensates of sorbitan monostearate with approximately 60 moles of ethylene oxide (e.g. those sold under the Trade Mark "TWEEN"), condensates of ethylene oxide or propylene oxide with propylene glycol (e.g. those sold under the Trade Mark "PLURONIC") and amphoteric agents such as quaternized imidazole derivatives which are available under the Trade Mark MIRANOL such as "Miranol C2M". Cationic surface active germicides and antibacterial compounds such as diisobutylphenoxyethyl dimethyl benzyl ammonium chloride, benzyl dimethyl stearyl ammonium chloride, tertiary amines having one fatty alkyl group (of from 12 to 18 carbon atoms) and two (poly) oxyethylene groups attached to the nitrogen (typically containing a total of from 2 to 50 ethanoxy

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groups per molecule) and salts thereof with acids, and compounds of the structure

(CH₂CH₂O)_zH (CH₂CH₂O)_xH -CH₂CH₂CH₂N

where R is a fatty alkyl group containing from 12 to 18 carbon atoms, and x, y and z total 3 or higher, as well as salts thereof with mineral or organic acids, may also be used. It is preferred to use from 0.05% to 5% by weight of surface-active material in the oral

preparations.

In certain forms of this invention a fluorineproviding compound is present in the oral preparation. These compounds may be slightly soluble in water or may be fully water-soluble. They are characterized by the ability to release fluoride ions in water and by substantial freedom from reaction with other compounds of the oral preparation. Among these materials are inorganic fluoride salts, such as suitable alkali metal, alkaline earth metal and heavy metal salts, for example, sodium fluoride, potassium fluoride, ammonium fluoride, lead fluoride, a copper fluoride such as cuprous fluoride, zinc fluoride, a tin fluoride such as stannic fluoride, or stannous chlorofluoride, barium fluoride, sodium fluorosilicate, ammonium fluorosilicate, sodium fluorozirconate, sodium monofluorophosphate, aluminium mono- and di-fluorophosphates, fluorinated sodium calcium pyrophosphate and sodium fluorozirconate. Alkali metal and tin fluorides, such as sodium and stannous fluorides and mixtures thereof, are preferred.

The amount of the fluorine-providing compound is dependent to some extent upon the type of compound, its solubility and the type of oral preparation, but it must be a nontoxic amount. In a solid oral preparation, such as a toothpaste or tooth powder, it is considered that an amount of such compound which releases a maximum of 1% by weight, based on the weight of the preparation, of fluoride ion is satisfactory. Any suitable minimum amount of such compound may be used but it is preferable to employ sufficient compound to release from 0.005% to 1%, must preferably about 0.1%, by weight of fluoride ion. Typically, in the cases of alkali metal fluoride and stannous fluoride, this component is present in an amount up to 2% by weight, based on the weight of the preparation, and preferably in the range from 0.05% to 1%. In the case of sodium monofluorophosphate the compound may be present in an amount up to 7.6% by weight, typically

In a liquid oral preparation such as a mouth wash, the fluorine-providing compound is typically present in amount sufficient to

release up to 0.13%, preferably 0.0013% to 0.1% and most preferably 0.0013% to 0.05%,

by weight of fluoride ions.

In certain forms of this invention an additional antibacterial agent in addition to the amino-cyano diguanide polymer may be present in amount to provide a total of about 5% by weight of antibacterial agent.. Such additional agents include:

 N^{1} - (4 - chlorobenzyl) - N^{5} - (2,4 -

dichlorobenzyl) biguanide; p-chlorophenyl biguanide;

4-chlorobenzhydryl biguanide; 4-chlorobenzhydrylguanylurea;

N - 3 - lauroxypropyl - N⁵ - p - chloro benzylbiguanide;

1,6-di-(p-chlorophenylbiguanidohexane); 1 - (lauryldimethylammonium) - 8 - (p chlorobenzyldimethylammonium)octane

chloride; 5,6-dichloro-2-guanidobenzimidazole;

N1-p-chlorophenyl-N5-laurylbiguanide; 5 - amino - 1,3 - bis(2 - ethylhexyl) - 5 -

methylhexahydropyramidine; and their non-toxic acid addition salts, particularly those where the anion includes a fluorine atom, such as; fluoride, monofluorophosphate, hexafluorosilicate and hexafluoro-aluminate. The dihydrogen fluoride is preferred, 1,6 - di - (p - chlorophenylbiguanido hexane) dihydrogen fluoride is particularly pre-

ferred.

The total amount of antibacterial agents in the oral preparation will generally be in the range 0.01% to 5% by weight and a major amount (i.e. more than 50% by weight) of the antibacterial agent is desirably the amino-cyano diguanide polymer, and the additional antibacterial agent is present in a minor

Various other materials may be incorpor- 100 ated in the oral preparations of this invention. Examples are colouring or whitening agents, preservatives, silicones, chloropyl compounds and ammoniated material such as urea, diammoniumphosphate and mixtures thereof. These adjuvants, where present, are incorporated in the preparations in amounts which do not substantially adversely affect the properties and characteristics desired.

Any suitable flavouring or sweetening 110 materials may also be employed. Examples of suitable flavouring constituents are flavouring oils, e.g. oils of spearmint, peppermint, wintergreen, sassafras, clove, sage, eucalyptus, marjoram, cinnamon, lemon and orange, and 115 sodiuim methylsalicylate. Suitable sweetening agents include sucrose, lactose, maltose, sorbitol, perillartine and saccharin. Suitably, flavour and sweetening agent may together comprise from 0.01% to 5% or more of the preparation.

In the manufacture of dentifrices, it is conventional to remove entrained air from the product by deaeration under vacuum, typically

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		at a late stage in the manufactur aspect of the present invention, it observed that in clear dentifrice gel	has been	EXAMPLE 2 Mouth-rinse	%	
5	5	able viscosity, the dispersed imm bubbles desirably enhance the appe- the dentifrice, and can, therefore be- to remain. Alternatively, the air	nobile air earance of permitted can be	Ethanol Glycerol Sodium monofluorophosphate Polymeric diguanide	4.0 1.0 0.2 0.5	65
1	0	replaced with another gas in non-to- tity, such as nitrogen or carbon di- particular, carbon dioxide can pre- effervescent character to the dentifric	ioxide. In rovide an	"Tween 80" Sodium saccharin Flavour/colour Water	0.5 0.2 1.0 To 100.0	70
		If it is desired to have a minimise of air in a dentrifrice of the prese	ed amount	EXAMPLE 3 Dentifrice		•
1	.5	tion, the "Unimix" apparatus des "Process Engineering", September pages 81—85, is particularly effica	scribed in 11, 1970, acious for	Hydrated alumina Glycerine Hydroxyethyl cellulose	% 55.0 20.0 1.0	75
		this purpose. In this apparatus a m can be rotated clockwise or counter		"Miranol C ₂ M"	1.0	80
2	0.	and the action of the mixing tool is by the action of a scraper blade that the working surface of the ap	to ensure paratus is	1,6-di-p-(Chlorophenylbiguanido- hexane) Polymeric diguanide Sodium saccharin	0.20 0.25 0.2	au
		scraped clean. Preferably a plastic such as polytetrafluorethane is use		Flayour	1.0	
2	5	scraper since it is compatible with tingredients of the dentifrice. The pof the mixing tool and the scrape	he various positioning	Water EXAMPLE 4 Dentifrice	To 100.0	85
4	30	raised central column in the apparat- further presence of a hydraulically vacuum-tight lid permits but litt	us and the operated le air to	Dicalcium phosphate dihydrate Calcium carbonate (precipitated) Glycerine	% 40.0 8.0 10.0	90
,	,,,	enter the composition during process the gelling agent and a portion of	the liquid	Sorbitol (70%) Sodium N-lauroyl sarcosinate	12.0 2.0	
		including water and/or humectant c ciently blended in the "Unimix"	apparatus.	Carboxymethylcellulose Sodium benzoate	0.9 0.2	95
3	35	Then the remaining liquid can be blended with the polishing agent	and addi-	Tetrasodium pyrophosphate Sodium saccharin	0.5 0.2	95
		tional components (except for post-ac ponents, such as flavouring oil) in	the "Uni-	Flavour Polymeric diguanide	1.0 0.5	
4	10	mix" apparatus, and then the two oblended together in the "Unimix" If desired, the small amount of a	apparatus. ir can be	Water EXAMPLE 5 Dentifrice	To 100.0	100
		largely removed under the depressu ditions in the apparatus. The appa be used to blend ingredients at r	aratus can oom tem-	Crystalline silica Polymethyl methacrylate	% 20.0	105
4	15	perature as well as at higher temper The following Examples illustrate		Glycerol Sodium lauroyl sulphate	20.0 20.0	105
		tion. The ingredients are mixed in manner. Percentages are by weight, meric diguanide employed is the	the usual The poly-	Hydroxyethyl cellulose Sodium fluoride	2.0 1.0 0.1	
5	50	polymeric product described above according to the procedure des	. prepared	Polymeric diguanide Sodium saccharin Flavour	0.5 0.2	110
		British Patent Specification No. 1,152,243.		Water	1.0 To 100.0	
		EXAMPLE 1		EXAMPLE 6 Dentifrice		115
		Mouth-rinse		Insoluble sodium metaphosphate	%· 45.0	
	55	Ethanol	% 15.0	Glycerol Sorbitol	6.0 16.0	
J.	در	"Pluronic 75" Sodium fluoride	0.5 0.05	Sodium N-lauroyl sarcosinate Sodium monofluorophosphate Polymeric diguanide	2.0 0.8 0.5	120
		Polymeric diguanide Sodium saccharin	0.1 0.2	Hydroxyethyl cellulose Sodium saccharin	1.0 0.2	
(60	Flavour/Colour Water	1.0 To 100.0	Flavour Water	1.0 To 100.0	125

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%

EXAMPLE 7 Dentifrice

			20.0
	Calcium carbonate (calcitic)		25.0
5	Galcium carbonate (aragonitic)		
_	Glycerol		20.0
	"Pluronic 75"		1.0
	Polymeric diguanide		0.25
	Polymeric diguamor		
	1,6-di-p-(Chlorophenyl biguanido-		
10	hexane)		0.20
10	Sodium saccharin		0.2
	Summi saccinatin		1.0
	Sodium carboxymethylcellulose		
	Water		100.0

WHAT WE CLAIM IS:-

1. An oral preparation comprising, as a non-toxic antibacterial agent, a polymer or a mixture of polymers having the formula:

wherein Y represents

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$$NH_2(CH_2)_6$$
—

OF

$$N \equiv C$$
— NH — C — NH — $(CH_2)_c$ — $\|$
 NH

X represents

$$-NH_2$$

25 or

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and n is an integer from 2 to 6,
or a salt or mixture of salts of such polymer or polymers; and, as a vehicle for the antibacterial agent, a mixture of water and alcohol or a solid water-insoluble dental polishing material.

An oral preparation as claimed in Claim
 wherein the polymer has the formula:

$$\begin{array}{c} \text{NH}_2\left(\text{CH}_2\right)_6 + \frac{1}{2} \text{NH} - \text{C} - \text{NH} - \text{C} - \text{NH} - \text{C} + \text{NH} - \text{C} - \text{NH} - \text{C}$$

3. An oral preparation as claimed in Claim 1 or Claim 2 wherein the antibacterial polymer is present in an amount in the range 0.01 to 5% by weight.

4. An oral preparation as claimed in any of the preceding Claims which is a mouth wash wherein the vehicle comprises a mixture of water and alcohol in a ratio in the range 1:1 to 20:1 by weight.

5. An oral preparation as claimed in any of Claims 1 to 3 which is a dentifrice wherein the vehicle comprises a solid water-insoluble polishing material.

6. An oral preparation as claimed in Claim 5 wherein the polishing material comprises insoluble sodium metaphosphate.

7. An oral preparation as claimed in Claim 5 wherein the polishing material comprises dicalcium phosphate (dihydrate or anhydrous).

8. An oral preparation as claimed in any of the preceding Claims which also contains a fluorine-providing compound.

9. An oral preparation as claimed in Claim 8 wherein the fluorine-providing compound comprises sodium fluoride.

10. An oral preparation as claimed in Claim 8 wherein the fluorine-providing compound comprises stannous fluoride.

11. An oral preparation as claimed in Claim 8 wherein the fluorine-providing compound comprises sodium monofluorophosphate.

12. An oral preparation as claimed in any of the preceding Claims which also contains a surface active agent.

13. An oral preparation as claimed in Claim 12 wherein the surface active agent is an anionic surface active agent.

14. An oral preparation as claimed in Claim 13 wherein the surface active agent comprises sodium lauryl sulphate.

15. An oral preparation as claimed in Claim 13 wherein the surface active agent comprises sodium N-lauroyl sarcosinate.

16. An oral preparation as claimed in any of the preceding claims which also contains an additional antibacterial agent in minor amount with respect to the antibacterial polymer, the total amount of antibacterial agents being in the range 0.01 to 5% by weight.

17. An oral preparation substantially as 85 described in any of the Examples.

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